

GRADE 11 PHYS PRE-TEST MARCH 2016 SEKHUKHUNE DISTRICT

QUESTION 1

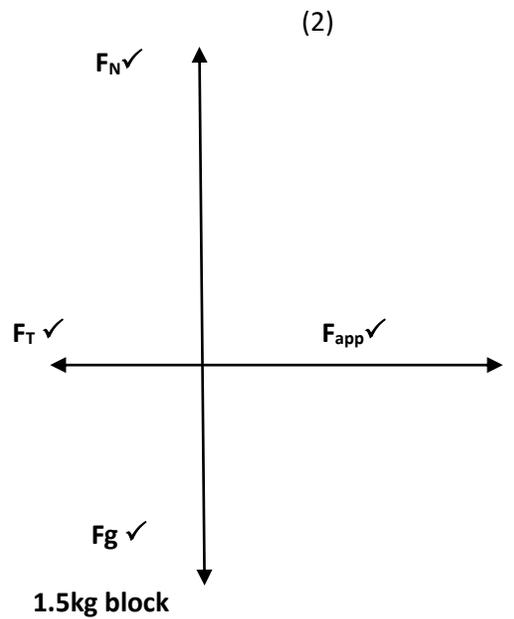
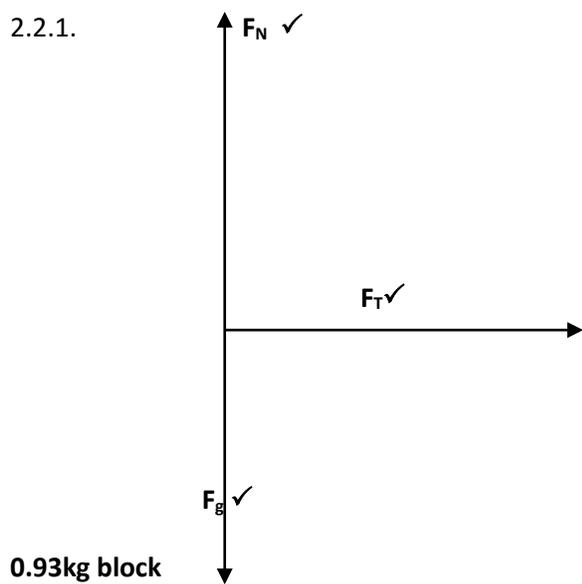
- 1.1. B. ✓✓ (2)
- 1.2. B. ✓✓ (2)
- 1.3. D. ✓✓ (2)
- 1.4. A. ✓✓ (2)
- 1.5. B. ✓✓ (2)
- 1.6. C. ✓✓ (2)
- 1.7. C. ✓✓ (2)
- 1.8. A. ✓✓ (2)
- 1.9. A. ✓✓ (2)
- 1.10. C. ✓✓ (2)

QUESTION 2

2.1 When a net force, F_{net} is applied to an object of mass, m , it accelerates in the direction of the net force. ✓ The acceleration, a , is directly proportional to the net force and inversely proportional to

the mass. ✓

2.2.1.



(2)

2.2.2. consider the 0.93kg block

$$F_{\text{net}} = m \times a = F_T \checkmark$$

$$0.93 \times a = F_T \checkmark \dots\dots\dots (1)$$

Consider the 1.5kg block

$$F_{\text{net}} = m \times a = F_{\text{app}} - F_T \checkmark$$

$$1.5 \times a \checkmark = 6.4 - F_T \checkmark \dots\dots\dots (2)$$

$$(2) + (1)$$

$$0.93 \times a + 1.5 \times a = 2.63 \text{m} \cdot \text{s}^{-2} \checkmark$$

Therefore $a = 2.63 \text{m} \cdot \text{s}^{-2} \checkmark$

2.2.3. consider the 0.93kg block

$$F_T = ma \checkmark$$

$$F_T = 0.93 = 2.63 \checkmark$$

$$F_T = 2.45 \text{N} \checkmark$$

2.2.4. decrease \checkmark

consider the 1.5kg block

$$m \times a = F_{\text{app}} - F_T \checkmark$$

$$1.5 \times 2.63 = 6.4 - F_T \checkmark$$

$$F_T = 2.45 \text{N} \checkmark$$

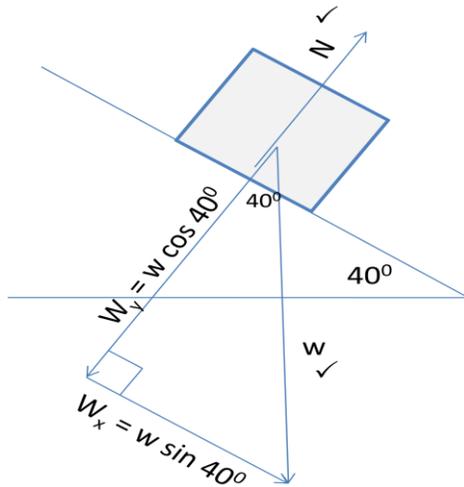
QUESTION 3

3.1. Static friction force : is the frictional force on an object standing still on a surface. \checkmark

Kinetic friction force: is the frictional force when an object moves with constant velocity over a surface. \checkmark (2)

3.2. The magnitude of the normal force = $w \cos \Theta$, so as the angle of the plane to the horizontal increases $\cos \Theta$ decreases and so the normal force decreases. \checkmark Because friction is proportional to the normal force a decrease in the normal force results in a decrease of friction. \checkmark (2)

3.3.1. $W = mg = 5 \times 9.8 = 49 \text{N}$; $\Theta = 40^\circ$



(2)

$$W_{\text{parallel}} = w \sin \theta \checkmark$$

$$= (49 \cdot \sin 40^\circ) \checkmark$$

$$= 31,50 \text{ N down the plane} \checkmark$$

(3)

$$W = 49; \theta = 40^\circ$$

$$W_{\text{perpendicular}} = W \cos \theta \checkmark$$

$$= 49 \cdot \cos 40^\circ \checkmark$$

$$= 37,54 \text{ N into the plane} \checkmark$$

(3)

3.3.2. For the block to start moving the component of its weight down the plane must be greater than the maximum static friction. \checkmark

(1)

$$N = 37,54 \text{ N}; \mu_s = 0,3; f_s^{\text{max}} = ?$$

(The magnitude of N equals the perpendicular component of the weight)

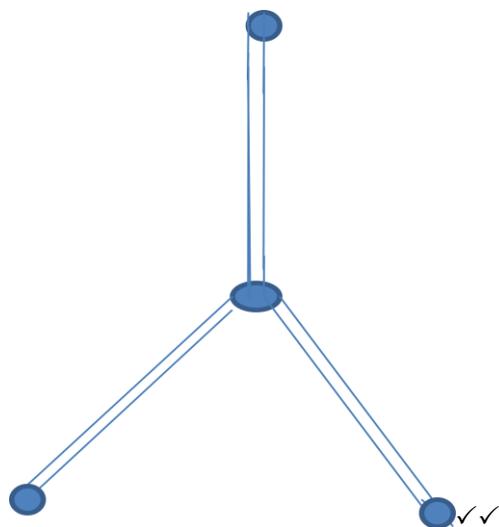
$$f_s^{\text{max}} = \mu_s \cdot N \checkmark$$

$$= 0,3 \cdot 37,54 \checkmark$$

$= 11,26 \text{ N} \checkmark$ the block will slide down the plane because f_s^{max} is less than the component of the weight (31,5 N) acting down the slope.

(3)

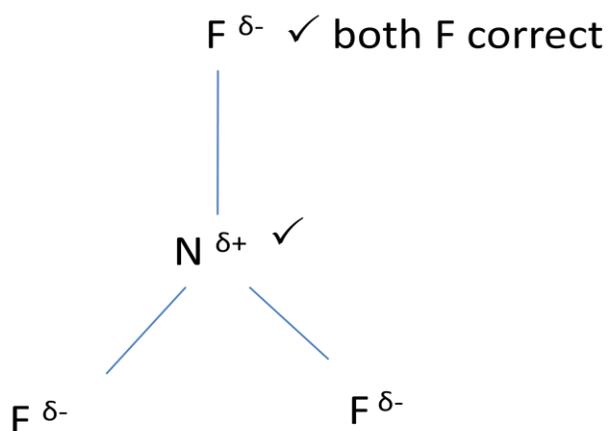
[16]



(for correct structure)

(2)

5.4. Fluorine atom is more electronegative, so has δ^- charge and nitrogen atom δ^+ .



(2)

5.5. NF_3 has a polar bond and is asymmetrical. ✓

(1)

[9]

QUESTION 6

6.1.1. 4 ✓

(1)

6.1.2. $-458 \text{ kJ}\cdot\text{mol}^{-1}$ ✓

(1)

6.1.3. $0,074 \text{ nm}$ ✓

(1)

6.1.4. The electronegativity difference between the two atoms is zero. ✓ This means that the electrons are shared equally. ✓ The electrons are shared by the overlapping of the two atomic orbitals to form a molecular orbital. ✓ A pure covalent bond forms. ✓

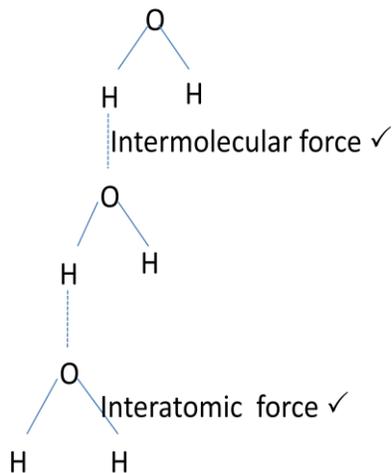
(4)

6.2.1. NH_3 = Dipole-dipole force/ dispersion forces ✓ (1)

H_2O = Dipole-dipole force/dispersion forces ✓ (1)

HF = Hydrogen bond ✓ (1)

6.2.2.



Intermolecular forces is an attraction between molecules. ✓ Interatomic forces is an electrostatic force of attraction between atoms due to sharing or transfer of electrons. ✓ Intermolecular forces are weaker than interatomic forces. (4)

6.2.3. Bent (angular) ✓ (1)

[15]

QUESTION 7

7.1. Ice cubes are less dense than liquid water. Water has a less dense solid phase than solid water. ✓ (1)

7.2. If the ice did not float on top of the water then all bodies of water would freeze from the bottom up. ✓ This would mean that aquatic life would not be able to survive through the cold winters as there would not be able to survive through the cold winters as there would be no habitat for them. ✓ (2)

7.3. High boiling point and low boiling point. ✓ Water has strong hydrogen bonds between molecules. These bonds require a lot of energy before they will break. This leads to water having a high boiling

point than if there were only weaker dipole-dipole forces. Water also has a high specific heat. ✓

If water did not have such a large range in which it is a liquid, bodies of water would freeze over faster, destroying the life in them. Also if the boiling point of water was lower, then all the water could evaporate on a hot day, which would cause all life to die. ✓ (3)

7.4. Water is able to absorb infrared radiation (heat) from the sun. ✓ This heat energy is stored in the oceans. ✓ Without this effect, the heat from the sun would cause the daytime temperatures on the Earth to become unbearably hot. ✓ (3)

[9]

TOTAL SECTION A [20]

TOTAL SECTION B [80]

GRAND TOTAL [100]